

## EFFICIENT CONSTRUCTION OF 6-AZASTEROIDS: DUAL INHIBITORS OF STEROIDAL 5 $\alpha$ -REDUCTASE

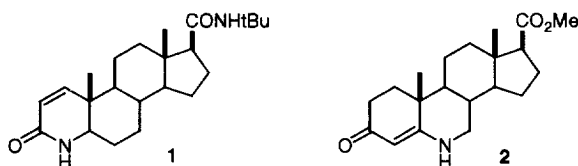
Matthew J. Sharp\* and Francis G. Fang

*Chemical Development Department, Glaxo Wellcome Inc, Research Triangle Park,  
North Carolina, 27709, U.S.A.*

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**Abstract:** A new route to 17 $\beta$ -substituted-6-azaandrost-4-en-3-ones, potent dual inhibitors of type 1 and 2 steroidal 5  $\alpha$ -reductase, is described. © 1998 Elsevier Science Ltd. All rights reserved.

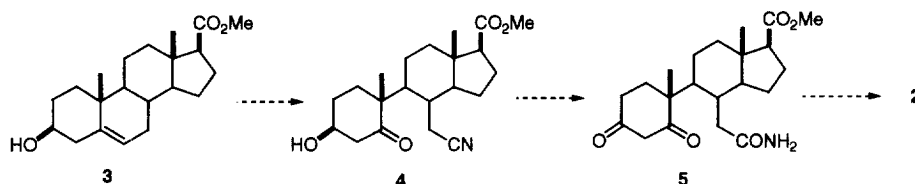
The use of steroidal 5 $\alpha$ -reductase (5AR) inhibitors for the treatment of certain androgen dependent conditions, such as benign prostatic hyperplasia (BPH), acne, male pattern baldness, and female hirsutism is now a well developed hypothesis.<sup>1</sup> In humans two isozymes of 5 $\alpha$ -reductase, types 1 and 2, catalyze the transformation of testosterone to dihydrotestosterone and are predominantly found in the skin and prostate, respectively.<sup>2</sup> Finasteride (**1**), a 4-azasteroid, was recently introduced as a type 2 selective 5AR inhibitor for the treatment of BPH.<sup>3</sup> In a search for potent dual inhibitors of 5AR, 6-azasteroids were recently reported by researchers at Glaxo Wellcome.<sup>4</sup> The prototypical compound in this series is the 17 $\beta$ -carbomethoxy-6-azaandrost-4-en-3-one **2**<sup>5</sup> shown below.



In order to fully evaluate the clinical potential of 6-azasteroids, such as **2**, a manufacturing route to these compounds was required. The previously reported synthesis of 17 $\beta$ -substituted-6-azaandrost-4-en-3-ones<sup>4</sup> starting from 3 $\beta$ -hydroxyetienic acid methyl ester<sup>6</sup> (**3**) followed the strategy reported by Lettre for 6-azacholesterol.<sup>7</sup> A significant drawback to this route in terms of safe large-scale production was the use of a Curtius rearrangement to introduce the nitrogen atom. Other mitigating circumstances included the use of several protecting groups and a number of chromium based oxidants. Finally, an oxidative cleavage of the C.5-C.6 olefin using ozone limited the throughput of the process. Because of problems associated with scale-up of the synthetic route described above it became necessary to explore a new route to 6-azasteroids and particularly to **2**.<sup>8</sup>

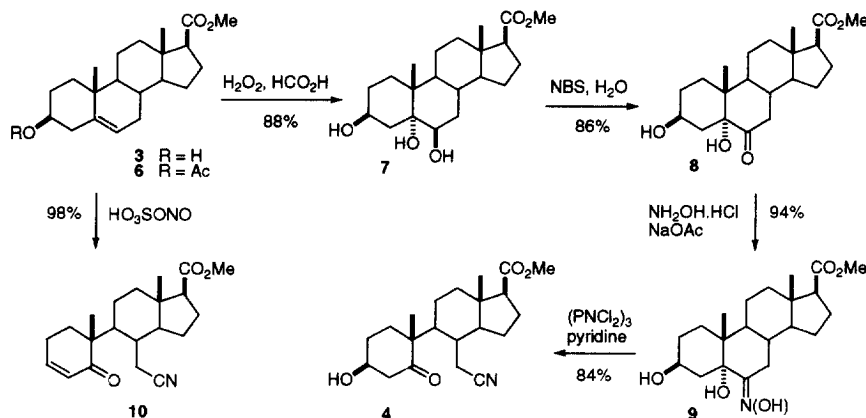
The synthetic plan for the preparation of **2** is set forth in Scheme 1. The starting material would continue to be derived from pregnenolone. A second-order Beckmann rearrangement was selected for cleavage of the C.5-C.6 bond.<sup>9,10</sup> Oxidation to a 1,3-diketone, hydrolysis of the nitrile to the corresponding amide, and Hofmann rearrangement<sup>11</sup> followed by cyclization would complete the synthesis of the 6-azasteroid. Herein is reported the successful realization of these goals and a synthesis of **2**.

Scheme 1



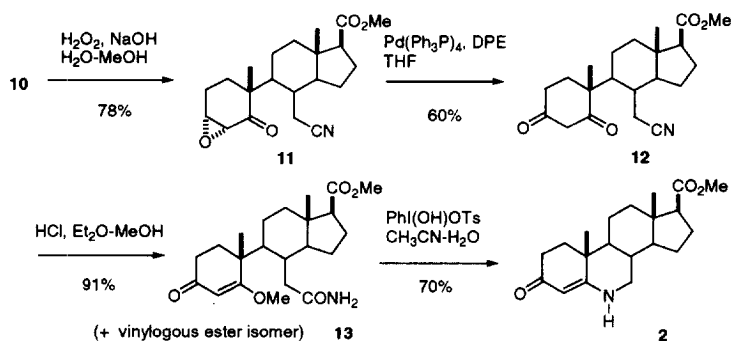
The first sub-goal, preparation of the  $\alpha$ -hydroxyoxime **9**, was realized in a straightforward fashion as outlined in Scheme 2. Oxidation of the  $\Delta^{5,6}$  olefin of  $3\beta$ -hydroxyetienic acid methyl ester (**3**) to the corresponding hydroxyketone **8** was accomplished using the two step sequence developed by Fieser.<sup>12</sup> Thus, **3** was treated with 96% formic acid and 30% hydrogen peroxide followed by hydrolysis with NaOH to afford the  $3\beta,5\alpha,6\beta$ -triol **7** in 88% yield. Regioselective oxidation of the  $6\beta$ -hydroxyl group with NBS in dioxane- $\text{H}_2\text{O}$  gave hydroxyketone **8** in 85% yield. Conversion to the corresponding  $\alpha$ -hydroxyoxime **9** was accomplished by heating **8** with hydroxylamine hydrochloride and sodium acetate in ethanol. The crystalline product **9** was isolated in 94% yield by filtration. The Beckman fragmentation of  $\alpha$ -hydroxyoxime **9** was first attempted using thionyl chloride as the activating agent. Under these conditions a 1:1 mixture of the 3-hydroxy 5,6-seconitrile **4** and the corresponding  $\beta$ -elimination product, enone **10** was obtained. Suppression of this side reaction was accomplished using milder activation conditions. Thus treatment of **9** with phosphonitrilic chloride trimer<sup>13</sup> in pyridine-THF afforded the 3-hydroxy 5,6-seconitrile **4** as the sole product in 84% yield. Overall, the 3-hydroxy 5,6-seconitrile **4** had been prepared in four steps and 60% yield overall from  $3\beta$ -hydroxyetienic acid methyl ester (**3**).

Scheme 2



A survey of the literature suggested a potentially more concise conversion of **3** to **4** by directly subjecting the C.5-C.6 olefin to nitrosation conditions.<sup>14</sup> In the event, treatment of 3 $\beta$ -hydroxyetienic acid methyl ester (**3**) with nitrosylsulfuric acid (2 equiv) in diethyl ether produced the enone 5,6-seconitrile **10** in 44% yield. Presumably the anticipated  $\alpha$ -hydroxyoxime **9** was formed as an intermediate that underwent the Beckmann fragmentation and a subsequent  $\beta$ -elimination to generate the observed product **10**. Repeating these conditions on the acetylated substrate **6** generated the enone 5,6-seconitrile **10** in 98% yield.<sup>15</sup> In view of the efficiency of this oxidative fragmentation process, we decided to investigate the plausibility of enone **10** as a precursor to the desired  $\beta$ -diketone moiety. Towards this end, the enone was subjected to epoxidation conditions (hydrogen peroxide, sodium hydroxide, methanol) as shown in Scheme 3. The  $\alpha$ -epoxide **11** crystallized directly from the reaction mixture in 78% yield<sup>16</sup> (ca. 10% of the  $\beta$ -epoxide remained in the filtrate). Rearrangement of the  $\alpha$ -epoxy-ketone **11** to the 1,3-diketone **12** was accomplished using the protocol of Noyori.<sup>17</sup> Thus, treatment of **11** with catalytic tetrakis(triphenylphosphine)palladium (0) and 1,2-bis(diphenylphosphino)ethane in refluxing tetrahydrofuran provided the 1,3-diketone **12** in 60% yield. Interestingly the minor  $\beta$ -epoxide was inert to these palladium catalyzed rearrangement conditions. With the steroid A-ring now at the correct oxidation state, the next step was to convert the nitrile to the corresponding primary amide. This hydrolysis was best performed by treating **12** with anhydrous HCl in Et<sub>2</sub>O/MeOH 3/1 at 0 °C followed by basic work-up. This reaction afforded the primary amide **13**, in which the 1,3-diketone had concurrently been transformed to a mixture of vinylogous methyl ester, in 91% yield.<sup>18</sup> With the primary amide in hand, the feasibility of the tandem Hofmann rearrangement/cyclocondensation reaction could be tested. This transformation was best accomplished using the conditions of Koser.<sup>19</sup> Thus treatment of the primary amide mixture **13** with 1.05 equiv of hydroxy(tosyloxy)iodobenzene in refluxing CH<sub>3</sub>CN/H<sub>2</sub>O 1/1 afforded the 6-azasteroid **2** in 70% yield.

Scheme 3



In summary, an efficient conversion of the readily available steroid **3** to 6-azasteroids such as the potent dual 5AR inhibitor **2** has been established (6 steps, 29% overall yield). Amongst the key features of this route are a one-step oxidative fragmentation of the C.5-C.6 double bond, (ii) a useful stereodependant application of the Noyori rearrangement to allow conversion of the enone to a 1,3-diketone, (iii) a tandem Hofmann rearrangement/cyclocondensation reaction to generate the 6-azasteroid skeleton. Furthermore, the above transformations avoid the use of ozone, protecting groups, heavy metal oxidants, and potentially hazardous azides.

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## References and Notes

1. For reviews, see: (a) Liang, T.; Rasmusson, G. H.; Brooks, J. R. *J. Steroid Biochem.* **1983**, *19*, 385. (b) Frye, S. V. *Curr. Pharm. Des.* **1996**, *2*, 59. (c) Kenny, B.; Ballard, S.; Blagg, J.; Fox, D. *J. Med. Chem.* **1997**, *40*, 1293.
2. Russell, D. W.; Wilson, J. D. *Annu. Rev. Biochem.* **1994**, *63*, 25.
3. Gormley, G. J.; Stoner, E.; Bruskewitz, R. C.; Imerato-McGinley, J.; Walsh, P. C.; McConnell, J. D.; Andriole, G. L.; Geller, J.; Bracken, B. R.; Tenover, J. S.; Vaughan, E. D.; Pappas, F.; Taylor, A.; Binkowitz, B.; Ng, J. *N. Engl. J. Med.* **1992**, *327*, 1185.
4. Frye, S. V.; Haffner, C. D.; Maloney, P. R.; Mook, R. A.; Dorsey, G. F.; Hiner, R. N.; Batchelor, K. W.; Bramson, H. N.; Stuart, J. D.; Schweiker, S. L.; van Arnold, J.; Bickett, D. M.; Moss, M. L.; Tian, G.; Unwalla, R. J.; Lee, F. W.; Tippin, T. K.; James, M. K.; Grizzle, M. K.; Long, J. E.; Schuster, S. V. *J. Med. Chem.*, **1993**, *36*, 4313.
5. Frye, S. V.; Haffner, C. D.; Maloney, P. R.; Hiner, R. N.; Dorsey, G. F.; Noe, R. A.; Unwalla, R. J.; Batchelor, K. W.; Bramson, H. N.; Stuart, J. D.; Schweiker, S. L.; van Arnold, J.; Bickett, D. M.; Moss, M. L.; Tian, G.; Lee, F. W.; Tippin, T. K.; James, M. K.; Grizzle, M. K.; Long, J. E.; Croom, D. K. *J. Med. Chem.* **1995**, *38*, 2621.
6. Rasmusson, S.; Reynolds, G. F.; Utne, T.; Jobson, R. B.; Primka, R. L.; Bergman, C.; Brooks, J. R. *J. Med. Chem.* **1984**, *27*, 1690.
7. Lettre, H.; Knof L. *Chem. Ber.* **1960**, *93*, 2860.
8. Fang, F. G.; Sharp, M. J. U.S. Patent 5 541 322, 1996; *Chem. Abstr.* **1996**, *125*, 86980.
9. Shoppee, C. W.; Roy, S. K. *J. Chem. Soc.* **1963**, 3775.
10. Ahmad, M. S.; Pillai, N. K.; Chaudhry, Z. H. *Aust. J. Chem.* **1974**, *27*, 1537.
11. A Hofmann rearrangement was the obvious choice for introduction of the nitrogen atom due to the well documented use of this reaction in industry. Wallace, E. S.; Lane, J. F. *Organic Reactions* **1946**, Vol III, 267.
12. Fieser, L. F.; Rajagopalan, S. *J. Am. Chem. Soc.* **1949**, *71*, 3938.
13. Rosini, G.; Medici, A. *Synthesis* **1975**, 665.
14. Narayanan, C. R.; Parker, M. S.; Wadia, M. S. *Tetrahedron Lett.* **1970**, *54*, 4703.
15. A similar transformation has been reported using Pb(OAc)<sub>4</sub> and TMSN<sub>3</sub> see Hugl, H.; Zbiral, E. *Tetrahedron* **1973**, *29*, 759.
16. The facial selectivity of the epoxidation was confirmed by X-ray crystallography.
17. Suzuki, M.; Watanabe, A.; Norori, R. *J. Am. Chem. Soc.* **1980**, *102*, 2095.
18. An isomeric mixture of vinylogous methyl esters was formed. These isomers could be separated and identified but for the purposes of the synthesis of **2** were carried forward as a mixture which converged at the target molecule.
19. Lazbin, I. M.; Koser, G. F. *J. Org. Chem.* **1986**, *51*, 2669.